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1. Compare protons, electrons, and neutrons with respect to charge, location, and mass. 2. What relationship exists between the mass number of an

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element and isotopes of that element? 3. Explain the difference between ionic and covalent bonds. 4. Compare and contrast adhesion and cohesion.

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Unit: Chemistry of life. 0. Legend (Opens a modal)
Possible mastery points. Skill Summary Legend
(Opens a modal) Structure of water and hydrogen
bonding. AP Bio: SYI (BI), SYI-1 (EU), Unit test. Level
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this set (29) A solution with a pH value of 12 is a
____. Base. Atoms are composed of ____.

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Click on the button next to the response that best
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(PDF) Chapter 02 Chemistry of Life Multiple Choice

Chapter 2 – Chemistry of Life Test Review 1. Review your notes, labs, and vocabulary for Chapter 2. 2. Review your basic chemistry worksheet! 3. Review the macromolecule and enzyme labs. 4. Make sure you know review all aspects of the McMush lab! Sugar – Benedict’s solution Lipid – Sudan III Starch – Lugol’s

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solution Proteins - Biurets 5.

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Catalase helps prevent a toxic buildup of hydrogen peroxide in cells by breaking it down into water and oxygen gas. Several students conduct an experiment to test the effects of pH on the activity of catalase. Each test tube contains a solution of hydrogen peroxide and water at various pH levels. The liver tissue is a source of catalase.

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